

The 7-Oxabicyclo[2.2.1]heptane System as a Valuable Starting Material for the Synthesis of Modified A Ring Vitamin D₃ Analogues

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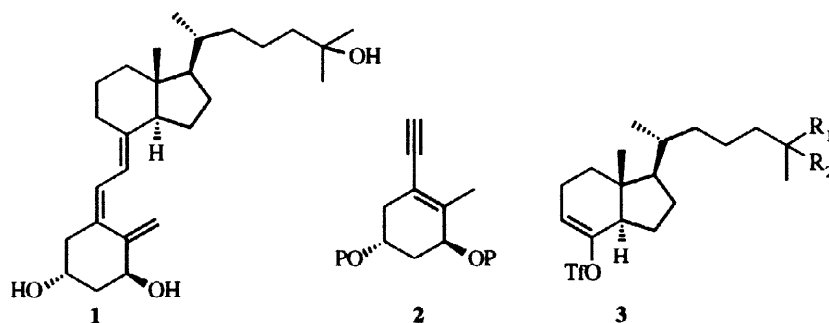
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Dedicated to Prof. Wolfgang Oppolzer, In Memoriam.

Abstract: A totally stereoselective synthesis of an analogue of the A ring fragment of vitamin D₃ has been achieved in optically pure form starting from a 7-oxanorbornenic disulfone. The key step is the alkylative cleavage of the oxygen bridge to produce a highly oxygenated cyclohexenyl sulfone. © 1998 Elsevier Science Ltd. All rights reserved.

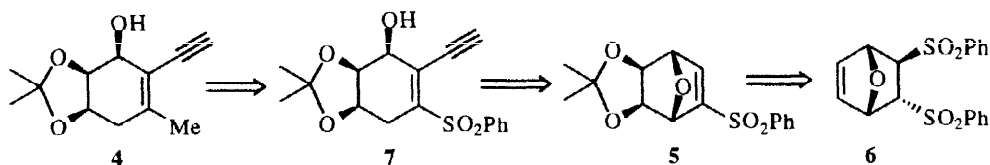
Since the discovery of the broad spectrum of biological properties of calcitriol¹ (1 α , 25-dihydroxyvitamin D₃ **1**) significant efforts have been made in order to develop novel vitamin D₃ analogues.² For these purposes, searches of versatile synthetic approaches to **1** and derivatives constitutes a main research objective. One of the most useful methods for the synthesis of **1** involves as a key step the Pd (II)-catalyzed coupling between enyne **2** (a precursor of the A ring of **1**) and a Grundmann's type enol triflate **3**³ (Scheme 1).



Scheme 1

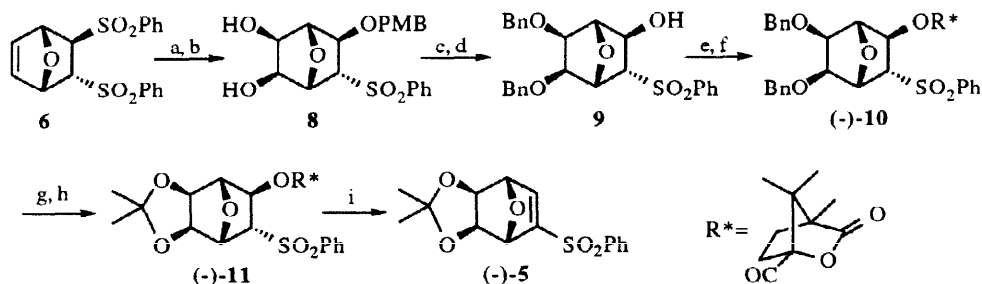
Although the synthesis of triflate **3** has been conveniently achieved from the commercially available vitamin D₂,⁴ a great deal of work has been done concerning the synthesis of compounds **2** and derivatives (and hence A ring derivatives).⁵ In this paper we wish to describe a new approach to these A ring analogues in optically pure form starting from 7-oxabicyclo[2.2.1]heptane derivatives.⁶ In our retrosynthetic approach for compound **4**, as a model for the evaluation of the method, we initially envisaged as a key step our methodology of alkylative ring opening⁷ of optically pure sulfone **5** obtained from compound **6** (the cycloadduct of furan and *E*-1,2-bis-(phenylsulfonyl)ethylene)⁸. The use of the corresponding lithium acetylide, followed by vinylsulfone

isomerization should give compound **7**. Final substitution of the sulfone functionality by a methyl group should afford the A ring analog **4** (Scheme 2). These predictions were later confirmed.



Scheme 2

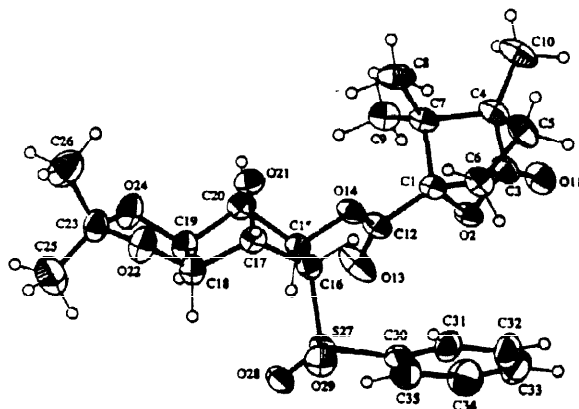
Optically pure vinylsulfone **5** was prepared from **6** according to the sequence summarized in Scheme 3.⁹



Key: a) PMBOH, KOH, CH₃CN;¹⁰ b) OsO₄, Me₃NO·2H₂O, acetone-H₂O 8:1, 90 % (two steps); c) NaH, BnBr, TBAI, 24 h.; 0 °C, THF; d) DDQ, CH₂Cl₂:H₂O 20:1, 24 h.; 82 % (two steps); e) R*Cl, Et₃N, CH₂Cl₂, 82 %, mixture of diastereomers; f) SiO₂ chromatography (hexane: AcOEt 2:1);¹¹ g) BF₃·OEt₂, EtSH; h) 2,2-dimethoxypropane, acetone, *p*-TsOH, 65 % (two steps), [α]²⁰_D = -41.0 (*c* = 1.0, CHCl₃); i) K₂CO₃, THF:H₂O 1:1, 71 %, [α]²⁰_D = -39.1 (*c* = 0.8, CHCl₃).

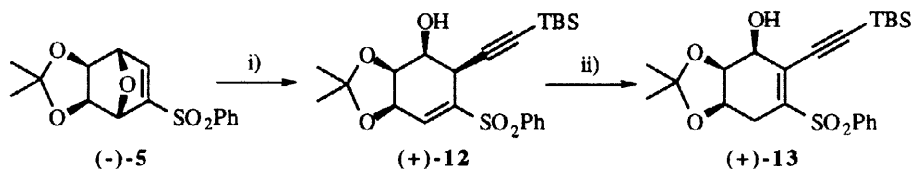
Scheme 3

The absolute configuration of ester **10** was determined by chemical correlation of the (+)-diastereomer with (+)-pinitol.¹² However, the vinylsulfone derived from **10** (analogous to **5** with benzyl as protecting groups) did not react in the conditions of the alkylative ring opening (conditions in Scheme 4, see below). Thus, the sequence was achieved with compound **11** obtained from **10** (see conditions in Scheme 3), and the absolute stereochemistry of the enantiomer (-)-**11** was confirmed by X-ray crystallography (Figure).¹³



Figure

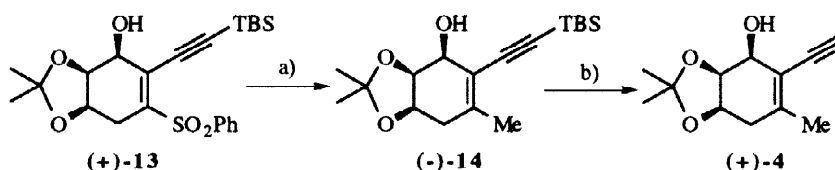
Operating at 0 °C in toluene, the alkylative oxa-bridge opening⁷ of (-)-5 by addition of lithium *tert*-butyldimethylsilylacetylide (generated from *tert*-butyldimethylsilylacetylene and *n*-BuLi in THF), produced cyclohexenylsulfone (+)-12 regio- and stereoselectively. The isomerization of (+)-12 to (+)-13 was performed using Na in MeOH at 0 °C or NaH in THF at 0 °C (Scheme 4).¹⁴



Key: i) *t*-butyldimethylsilylacetylene, *n*-BuLi, THF:toluene, 0 °C, 62 %, $[\alpha]_{D}^{20} = +0.05$ ($c = 3.7$, CHCl₃). ii) Na/MeOH, 0 °C, 71 %, $[\alpha]_{D}^{20} = +265.0$ ($c = 1.2$, CHCl₃).

Scheme 4

The substitution of the phenylsulfonyl functionality by a methyl group in (+)-13 was carried out with MeMgBr and catalytic amounts of Ni(acac)₂ in THF¹⁵ to give (-)-14 in 73 % yield, which was transformed into (+)-4 by reaction with TBAF in THF at room temperature (Scheme 5).



Key: a) MeMgBr, Ni(acac)₂, THF, 73 %, $[\alpha]_{D}^{20} = -33.5$ ($c = 0.2$, CHCl₃). b) TBAF, THF, 80 %, $[\alpha]_{D}^{20} = +34.1$ ($c = 0.4$, CHCl₃).

Scheme 5

In summary, in this report a new efficient methodology for the synthesis of A ring analogues of vitamin D₃ in optically pure form has been developed. Applications of this procedure for the synthesis of derivatives of 1 functionalized in the A ring are now under active research in our laboratory.

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References and notes.

- #. To whom the correspondence related to the X-ray determination should be addressed.
1. See, for instance, Norman, A. W.; Bouillon, R.; Thomasset, M. in "Vitamin D, A Pluripotent Steroid Hormone: Structural Studies, Molecular Endocrinology and Clinical Applications". Walter de Gruyter, Berlin, 1994.
2. For two excellent reviews on the chemistry and structure of Vitamin D₃ and analogues, see: a) Zhu, G.; Okamura, W. H.; *Chem. Rev.* **1995**, *95*, 1877-1952. b) Okamura, W. H.; Midland, M. M.; Hammond, M. W.; Rahman, M. A.; Dormanen, M. C.; Nemere, I.; Norman, A. W.; *J. Steroid Biochem. Molec. Biol.* **1995**, *53*, 603-613. For some recent references on the synthesis and chemistry of analogues of Vitamin D₃, see: c) Dauben, W. G.; Zhou, B.; Law, J. Y. L.; *J. Org. Chem.* **1997**, *62*, 9005-9008. d) Torneiro, M.; Fall, Y.; Castedo, L.; Mouriño, A.; *Tetrahedron* **1997**, *53*, 10851-10870.; e) Posner, G. H.; Lee, J. K.; White, M. C.; Hutchings, R. H.; Dai, H.; Kachunski, J. L.; Dolan, P.; Kensler, T. W.; *J. Org. Chem.* **1997**, *62*, 3299-3314. f) Fall, Y.; *Tetrahedron Lett.* **1997**, *38*, 4909-4912. g) Oshida, J.;

- Okamoto, M.; Azuma, S.; Tanaka, T.; *Tetrahedron: Asymm.* **1997**, *8*, 2579-2584. h) Grzywacz, P.; Marczak, S.; Wicha, J.; *J. Org. Chem.* **1997**, *62*, 5293-5298. i) Fall, Y.; Castedo, L.; Mouriño, A.; *J. Org. Chem.* **1997**, *62*, 6344-6352. j) García, A. M.; Mascareñas, J. L.; Castedo, L.; Mouriño, A.; *J. Org. Chem.* **1997**, *62*, 6353-6358.
3. a) Castedo, L.; Mouriño, A.; Sarandeses, L. A.; *Tetrahedron Lett.* **1986**, *27*, 1253-1256. b) Castedo, L.; Mascareñas, J. L.; Mouriño, A.; Sarandeses, L. A.; *Tetrahedron Lett.* **1988**, *29*, 1203-1206. c) Castedo, L.; Mascareñas, J. L.; Mouriño, A.; Sarandeses, L. A.; *Tetrahedron* **1991**, *47*, 3485-3498.
 4. a) Enas, J. D.; Shen, G. Y.; Okamura, W. H.; *J. Am. Chem. Soc.* **1991**, *113*, 3873-3881. b) Curtin, M. L.; Okamura, W. H.; *J. Am. Chem. Soc.* **1991**, *113*, 6958-6966. For some more recent approaches, see, for instance: c) Michalak, K.; Stepanenko, W.; Wicha, J.; *Tetrahedron Lett.* **1996**, *37*, 7657-7658. d) Clasby, M. C.; Craig, D.; Jaxa-Chamiec, A. A.; Lai, J. Y. Q.; Marsh, A.; Slawin, A. M. Z.; White, A. J. P.; Williams, D. J.; *Tetrahedron* **1996**, *52*, 4769-4802.
 5. For an account of the different methodologies, see ref. 2a., pp. 1905-1907. For some recent references, see: a) Fall, Y.; Torneiro, M.; Castedo, L.; Mouriño, A.; *Tetrahedron* **1997**, *53*, 4703-4714. b) Parker, K. A.; Dermatakis, A.; *J. Org. Chem.* **1997**, *62*, 6692-6696. c) Ferreira, M.; Fernández, S.; Gotor, V.; *J. Org. Chem.* **1997**, *62*, 4358-4363. d) Mouriño, A.; Torneiro, M.; Vitale, C.; Fernández, S.; Pérez-Sestelo, J.; Anné, S.; Gregorio, G.; *Tetrahedron Lett.* **1997**, *38*, 4713-4716. e) Konno, K.; Maki, S.; Fijishima, T.; Liu, Z.; Miura, D.; Chakki, M.; Takayama, H.; *Bioorganic Med. Chem. Lett.* **1998**, *8*, 151-156.
 6. For other approaches to A ring precursors starting from a 7-oxabicyclo[2.2.1]heptane derivative, see, De Schrijver, J.; De Clercq, P. J.; *Tetrahedron Lett.* **1993**, *34*, 4369-4372.
 7. Arjona, O.; de Dios, A.; Fernández de la Pradilla, R.; Plumet, J.; Viso, A.; *J. Org. Chem.* **1994**, *59*, 3906-3916. b) Aceña, J. L.; Arjona, O.; Iradier, F.; Plumet, J.; *Tetrahedron Lett.* **1996**, *37*, 105-106.
 8. De Lucchi, O.; Lucchini, V.; Pasquato, L.; Modena, G.; *J. Org. Chem.* **1984**, *49*, 596-604.
 9. All new compounds showed spectroscopic and analytical data consistent with the assigned structures. In the case of ^1H NMR, selective decoupling experiments were used for stereochemical determinations. Full details will be published elsewhere.
 10. Mirsadeghi, S.; Rickborn, B.; *J. Org. Chem.* **1985**, *50*, 4340-4345.
 11. The sequence was performed with both diastereomers. Specific rotations are indicated for the products obtained from diastereomer (-)-**10**.
 12. Aceña, J. L.; Arjona, O.; Plumet, J.; *Tetrahedron: Asymm.* **1996**, *7*, 3535-3544.
 13. Crystal data: (-)-**11**, $\text{C}_{25}\text{H}_{30}\text{O}_9\text{S}$, $M = 506.567$, monoclinic, space group P21, $a = 15.2880$ (0.0010); $b = 13.9790$ (0.0010); $c = 6.0190$ (0.0010) Å; $\alpha = \gamma = 90.00$; $\beta = 94.50$; $V = 1282.4$ (0.3) Å³; $Z = 2.00$; $d_c = 1.312$ Mg.m⁻³, $F(000) = 536$. 2599 independent reflections were collected on four circle diffractometer Phillips PW 1.100. The structure was determined by direct methods and refined on L. S. on Fobs. H atoms parameters not refined. Coordinates have been deposited at the Cambridge Crystallographic Data Centre.
 14. See for a different process depending of the silyl protecting group nature, Arjona, O.; Borrallo, C.; Iradier, F.; Medel, R.; Plumet, J.; *Tetrahedron Lett.* **1998**, *39*, 1977-1980.
 15. Fabre, J. L.; Julia, M.; Verpeaux, J. N.; *Tetrahedron Lett.* **1982**, *23*, 2469-2472.